#### REMARKS

Claims 1-17 are pending in the application.
Claim 17 has been subjected to a restriction requirement.
Claims 11 and 12 have been cancelled by this amendment.
Accordingly, claims 1-10 and 13-17 are at issue.

The courteous interview granted by Examiner Ngo to applicant's undersigned attorney on January 14, 1998 is hereby acknowledged with appreciation. During the interview, the Office Action and proposed claim amendments were discussed.

With respect to the Office Action, it was stated that compounds of formula (II) of claim 17 are related to the compounds and process of the claims of elected Group I, and that the compounds of formula (II) would be considered as a part of Group I. Accordingly, applicant has amended claim 17 by deleting the compounds of formulae (III), (V), (VI), (VII), (VIII), and (X). Claim 17, therefore, should be considered on the merits at this time. In addition, claim 16 has been amended to delete processes (B) and (C) from claim 16.

The Office Action indicates that applicant has not filed a certified copy of GB 9401090.7, which applicant relies upon for a claim of foreign priority. However, it is submitted that applicant is not required to submit a certified copy of the priority document in this case.

The present application is the U.S. national phase application of PCT/EP95/00183. Accordingly, it is applicant's understanding that under the procedures of the PCT, a copy of the priority document will have been supplied to the U.S. Patent Office, pursuant to Rule 17 of the PCT regulations. Accordingly, it is requested that the next communication concerning this application

contains an indication that the appropriate priority document is in the file of this application.

The Office Action also requests that a reference to an earlier-filed, and copending, application should be added to the present application. It is submitted that such a reference is not needed because there was no previously filed application in the U.S. A PCT application, designating the U.S., was filed on January 19, 1995. Filing of the PCT application in the U.S. was perfected by entering the U.S. national phase on July 17, 1996. Accordingly, no reference to a previously filed and copending application is necessary.

Claims 1-16(A) stand rejected under 35 U.S.C. \$112, first and second paragraphs, as being nonenabling or indefinite. In addition, claims 11 and 12 stand rejected under 35 U.S.C. \$101. In view of the amendments to the claims, and for the reasons set forth below, it is submitted that these rejections have been overcome and should be withdrawn.

In particular, it is contended that the phrase " $R^1$  and  $R^3$  together represent . . . alkyl or alkenyl chain" is not clear. As indicated in the Office Action, this phrase is intended to recite that  $R^1$  and  $R^3$  are taken together to form a ring. Accordingly, claim 1 has been amended to recite that  $R^1$  and  $R^3$  are taken together as a component of a 5- or 6-membered ring.  $R^1$  and  $R^3$ , together, contribute three or four members of the ring, and the remaining two members are the nitrogen atom bonded to  $R^1$  and the carbon atom bonded to  $R^3$ .

From a reading of all substituents  $R^1$  and  $R^3$  recited in claim 1, the phrase in question can only be construed as taking  $R^1$  and  $R^3$  together to form a ring. For example,  $R^1$  can be  $C_{1-6}$  alkyl,  $C_{2-6}$  alkenyl, or  $C_{2-6}$  alkynyl (and others), and  $R^3$  can be hydrogen or  $C_{1-3}$  alkyl.

Based on those definitions of  $R^1$  and  $R^3$ , the only reasonable construction of taking  $R^1$  and  $R^3$  together is to form a ring. Any other construction would be redundant in view of the recited definitions of  $R^1$  and  $R^3$ . It is submitted, therefore, that the amendment to claim 1 overcomes this rejection under 35 U.S.C. §112, second paragraph.

The phrase "and salts and solvates" also is considered indefinite. Accordingly, applicant has amended claims 1, 2, 8, 9, and 10 to recite "or salts or solvates." This amendment clarifies that there are not multiple forms of the compound of formula (I) in one compound. It again is submitted that these amendments to claims 1, 2, 8, 9 and 10 overcome this rejection under 35 U.S.C. §112, second paragraph.

Claims 11 and 12 stand rejected under 35 U.S.C. \$112, second paragraph, and 35 U.S.C. \$101. In view of this amendment, which cancels claims 11 and 12, it is submitted that these rejections are now moot.

Claims 1-16(A) stand rejected under 35 U.S.C. \$112, first paragraph, as not being enabling for a compound of formula (I) when R<sup>1</sup> and R<sup>3</sup> are taken together form a chain, because such a compound cannot be synthesized. It is submitted that this contention is incorrect, and, for the reasons set forth below, this rejection should be withdrawn.

As stated above, when  $R^1$  and  $R^3$  are taken together, these substituents form a ring. To illustrate how a compound of formula (I) can be made when  $R^1$  and  $R^3$  are taken together to form a ring, the examiner's attention is directed to compound (IX) at page 13 of the specification. As stated in the specification,

"There is further provided by the present invention a process (B) for preparing a compound of formula (I), wherein R¹ and R³ together represent a 3- or 4-membered alkyl or alkenyl chain, which process (B) comprises cyclisation of a compound of formula (VIII)" (page 13, lines 3-6), and

"Conveniently a compound of formula (VIII) is prepared by reaction of a compound of formula (III) as hereinbefore described with a compound of formula (IX)" (page 13, lines 13-15).

The compound of formula (III) is illustrated in the specification at the top of page 10.

Therefore, to prepare a compound of formula (I) wherein  $R^1$  and  $R^3$  are taken together to form a ring, the following reaction sequence is used.

The compound of formula (III) is disclosed at page 10 of the specification. The cyclization step is disclosed at page 13, lines 8-12.

The synthesis of a compound of formula (I), wherein  $R^1$  and  $R^3$  are taken together to form a ring, therefore, can be accomplished by providing a compound of formula (IX), wherein  $R^1$  and  $R_3$  are taken together to form a ring. In particular, a compound of formula (IX) has the following structure.

$$\begin{array}{c|c} O & \mathbb{R}^4 \\ \parallel & \mid \\ \text{Hal-C-C-N-R}^1 \\ & \mid \\ \mathbb{R}^3 \end{array}$$

A compound of formula (IX) having  $R^1$  and  $R^3$  taken together to form a 5- or 6-membered ring could have the structure

wherein Hal represent a halogen atom, like chlorine (see specification, page 9, lines 26-27), and R<sup>4</sup> is a protecting group, like benzyloxycarbonyl (see specification, page 13, lines 17-18). Synthesis of the above compound would provide a compound of formula (IX), which in turn could be used to prepare a compound of formula (I) by the above reaction sequence.

The above compound can be prepared from compounds and synthetic steps well known to persons skilled in the art. For example, a suitable starting material would be picolinic acid or proline, having the following structures (a) and (b), respectively.

These compounds are available commercially, as illustrated in pages 1019 and 1057, of the 1992 catalog of

Aldrich Chemical Co., Milwaukee, WI, attached hereto as Exhibit A. In these starting materials, R<sup>1</sup> and R<sup>3</sup> are taken together to form a 4-membered alkyl chain and a 3-membered alkyl chain, respectively, as a component of a 6-membered ring and a 5-membered ring, respectively.

Compound (a) or (b) then can be reacted with a protecting compound, like benzylchloroformate, to position a protecting group, like benzyloxycarbonyl, on the nitrogen atom. This reaction is illustrated in C.D. Gutsche et al., "Fundamentals of Organic Chemistry," Prentice-Hall, Inc., Englewood Cliffs, NJ, page 1190, attached hereto as Exhibit B. This reaction is illustrated below for compound (a) and an identical reaction can be performed on compound (b).

Benzylchloroformate is available commercially, as illustrated at page 132 of Exhibit A.

Compound (c) then can be reacted with a common reagent for converting a carboxylic acid to an acid chloride, like thionyl chloride ( $SOCl_2$ ), to provide com-

pound (d). This reaction is illustrated at page 39 of Exhibit B.

$$\begin{array}{c|c}
 & \text{SOCl}_2 \\
 & \text{N} & \text{C-Cl} \\
 & \text{R}^4
\end{array}$$
(c) (d)

Compound (d) corresponds to the compound of formula (IX) at page 13 of the specification, wherein Hal is chlorine,  $R^4$  is benzyloxycarbonyl, and  $R^1$  and  $R^3$  are taken together as a 4-membered alkyl chain component of a 6-membered ring. An identical reaction sequence starting with proline would yield an identical compound of formula (IX), except  $R^1$  and  $R^3$  are taken together as a 3-membered alkyl component of a 5-membered ring.

Compound (d), or a similar compound prepared from proline, then could be reacted with a compound of formula (III) to yield a compound of formula (VIII), which in turn is cyclized to form a compound of formula (I).

Therefore, a compound of formula (I) can be prepared when  $R^1$  and  $R^3$  are taken together as a 3- or 4-membered alkyl or alkenyl chain. The synthesis utilizes well-known starting materials, reagents, and reactions. Accordingly, it is submitted that the rejection of claims 1-16(A) under 35 U.S.C. §112, first paragraph, should be withdrawn.

During the interview, the examiner stated that the claims appeared excessive in scope because of the terms aryl and heteroaryl. In response, applicant has amended claim 1 to more particularly claim the aryl and heteroaryl substituents. Support for this amendment can be found in the specification at page 2, lines 16-17.

It is submitted that the claims are now in proper form and scope for allowance. Early and favorable action on the merits are respectfully requested.

Should the examiner wish to discuss the foregoing, or any matter of form in an effort to advance this application toward allowance, the examiner is urged to telephone the undersigned at the indicated number.

Respectfully submitted,

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Ву

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Benzylbis(triphenylphosphine)paliadium(II) chloride, see 27,766-5, trans-Benzyl- (chloro)bis(triphenylphosphine)paliadium(III) page 132   132   132   132   134
mp 3 to -1° bp 198-199° ng 15750 d 1.438 Fp 188°F(86°C) Bell. 5,306   100g 23.80
mp 3 to -1° bp 198-199° ng 15750 d 1.438 Fp 188°F(86°C) Bell. 5,306   100g 23.80
mp 3 to -1° bp 198-199° ng 15750 d 1.438 Fp 188°F(86°C) Bell. 5,306   100g 23.80
24,563-1 Benzyl 2-bromoacetate, 96% [5437-45-6] BrCH,CO,CH,C,H, FW 229.08
24,563-1 Benzyl 2-bromoacetate, 96% [5437-45-6] BrCH,CO,CH,C,H, FW 229.08
bp 166-170°/22mm nB 1.5440 d 1.446 Fp > 230°F(110°C) Bell. 6(1).220 250g 15.00 18A:04 1.446 Fp > 230°F(110°C) Bell. 6(1).220 250g 15.00 18A:04 1.446 Fp > 230°F(110°C) Bell. 6(1).220 250g 15.00 18A:04 1.446 Fp > 230°F(110°C) Bell. 6(1).220 250g 15.00 18A:04 18A
NMR 2(2),269B   FT-IR 1(3),1336D   Safety 2,381C   RTECS# AF5957215   Disp. A
38,204-3 Benzyl 3-bromopropyl ether [54314-84-0] C <sub>4</sub> H <sub>4</sub> CH <sub>2</sub> O(CH <sub>2</sub> ) <sub>3</sub> Br FW 229.12
38,204-3 Benzyl 3-bromopropyl ether [54314-84-0] C <sub>4</sub> H <sub>4</sub> CH <sub>2</sub> O(CH <sub>2</sub> ) <sub>3</sub> Br FW 229.12
Disp. A IRRITANT Benzyl-tert-butanol, see B1,800-6, α,α-Dimethylbenzenepropanol page 492  **EW 312.37 nb 1.5400 d 1.100 Fp > 230°F(110°C) Bell. 9(2),594  **B1,820-0 B1,820-0 Disp. A IRRITANT  Disp. A IRRITANT  19 9.00 62.
### Suppose
Benzyl butyl phthalate, 98% [85-68-7] 2-{CH <sub>s</sub> (CH <sub>s</sub> ) <sub>2</sub> O <sub>2</sub> C C <sub>s</sub> H <sub>s</sub> CO <sub>2</sub> CH <sub>s</sub> CO <sub>3</sub> CH <sub>s</sub> CO <sub></sub>
F7-IR 1(3),1377A Safety 2,381D R7ECS# TH9990000 Disp. A IRRITANT 10.00 250mi 14.60
FT-IR 1(3),1377A Safety 2,381D RTECS# TH9990000 Disp. A IRRITANT 10.00 250mi 14.60
Disp. A //RRITANT 250ml 14.60
Disp. A //RRITANT 250ml 14.60
Bell. 6,437 NMR 2(2),361A FT/B HMNCO-CH <sub>2</sub> C <sub>2</sub> H <sub>6</sub> FW 151.17 mp 97.000
12,101-0 S-Benzyl-N-carbobenzyloxy-t-cystelpo 2004 Disp. A 25g 21.20
C.H.CH SCH CHARLES TO STORE 98% 12257 40 65
[a] <sup>21</sup> 44° (c = 2, C, H <sub>0</sub> OH) NMA 2(2) 264C FT 10.2 mp 94-96° 109 13.05
(*DB07VL3.46-bases 1
1-benzyl-4-oxo-3-piperidone hydrochloride, see 22,700-5, Methyl  22,900-8 Benzylcetyldimethylammonium chloride monohydrate  C <sub>4</sub> H <sub>2</sub> CH <sub>3</sub> (CH <sub>3</sub> ) <sub>4</sub> CH <sub>3</sub> (CH <sub>3</sub> ) <sub>5</sub> CH <sub>3</sub> (CH <sub>3</sub> ) <sub>6</sub> CH <sub>3</sub> (CH <sub>3</sub> ) <sub>6</sub> CH <sub>3</sub> (CH <sub>3</sub> ) <sub>6</sub> CH <sub>3</sub> (CH <sub>3</sub> (CH <sub>3</sub> ) <sub>6</sub> CH <sub>3</sub> (CH <sub>3</sub> ) <sub>6</sub> CH <sub>3</sub> (CH <sub>3</sub> (CH <sub>3</sub> ) <sub>6</sub> CH <sub>3</sub> (CH <sub>3</sub> ) <sub>6</sub> CH <sub>3</sub> (CH <sub>3</sub> (CH <sub>3</sub> ) <sub>6</sub> CH <sub>3</sub> (CH <sub>3</sub> ) <sub>6</sub> CH <sub>3</sub> (CH <sub>3</sub> ) <sub>6</sub> CH <sub>3</sub> (CH <sub></sub>
Merck Index 11,2009 NMB 2(1) 1130 PM 414.12 mp 62.64° Bell 12(3) 2010 5g 8.80
Merck Index 11,2009 NMR 2(1),1122B FT-IR 1(1),1321C Safety 2,382A 13.20  28,847-0 Benzyl-α-12C Chloride 90
28,847-0 Benzyl-α- <sup>12</sup> C chloride, 99 atom % <sup>12</sup> C [57742-41-3] (α-chlorotoluene-α- <sup>12</sup> C)
C <sub>6</sub> H <sub>2</sub> <sup>13</sup> CH <sub>2</sub> Cl FW 127.58 mp 43° bp 177-181° nb 1.5380 d 1.100 38.70 CANCER SUSPECT ACCUSE.
Fp 165°F(73°C) Safety 2,382C Disp. C HIGHLY TOXIC 19 455.80 (Packaget) 43 455.80
IN SUSPECT AGENT TOXIC 19 455 80 1
(Packaged in prescored ampules)  21,733-6 Benzyl-d, chloride, 99 + atom % D [59502-05-5] (α-chlorotoluene-d,)
Benzyl-d, chloride, 99 + atom % D 150500 00 -
C <sub>2</sub> D <sub>2</sub> CD <sub>2</sub> Cl FW 133.64 bb 65°(10mm   199502-05-5] (α-chlorotoluene-d)
C <sub>c</sub> D <sub>c</sub> CD <sub>c</sub> Cl FW 133.64 bp 65°/10mm n <sub>B</sub> 1.5374 d 1.200 Fp 165°F(73°C)  NMR 2(1),6C FT-IR 1(3),1616C Safety 2,383A Disp. C HIGHLY 70°C)  18 FF 2 CANCER SUSPECT AGENT
NMR 2(1),6C FT-IR 1(3),1616C Safety 2,383A Disp. C HIGHLY TOXIC  18,555-8  8enzyl chloride, 99% [100-44-7] (α-chlorotoluene) C <sub>2</sub> H <sub>2</sub> CH <sub>2</sub> Cl FW 126.59
Bell2y Chloride, 99% [100-44-7] (archloretal
* mp 43° bp 177-181° ng 1.5380 d 1.100 Fp 165°F(73°C) Bell. 5,292 Merck 250g 12.90  ** MTECS# XS8925000 Disp. C Highly Cross-200 A Safety 2,3828 Merck 250g 12.90
RIFCS# VSPORSED 509 10.00 Fp 165°F(73°C) Bell. 5.292 Merch 509 10.00
Mutagen, possible carcinogen, Proc. Net. ASSE 250g 12.90 Inhibited 17.90
Mutagen, possible carcinogen, Proc. Not. No. No. No. No. No. No. No. No. No. No
Mutagen, possible carcinogen, Proc. Not. No. No. No. No. No. No. No. No. No. No
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).  32,016-1  Benzyl chloride, 97% [100-44-7] (α-chlorotoluene) C <sub>8</sub> H <sub>1</sub> CH <sub>2</sub> Cl
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).  32,016-1  Benzyl chloride, 97% [100-44-7] (α-chlorotoluene) C <sub>8</sub> H <sub>1</sub> CH <sub>2</sub> Cl
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).  32,016-1  Benzyl chloride, 97% [100-44-7] (α-chlorotoluene) C <sub>8</sub> H <sub>1</sub> CH <sub>2</sub> Cl
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).  Salety 2,3328  Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).  Senzyl chloride, 97% [100-44-7] (a-chlorotoluene) C <sub>4</sub> H <sub>4</sub> CH <sub>4</sub> CI
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).  32,016-1  **    Senzyl chloride, 97% [100-44-7] (\(\alpha\)-chlorotoluene)   C <sub>4</sub> H <sub>4</sub> CH <sub>2</sub> CI
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).   12.90   12.90   12.90   14.90   17.90   17.90   18.90   17.90   18.90
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).   12.90   12.90   12.90   14.90   17.90   17.90   18.90   17.90   18.90
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).   1kg   17.90
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).   1kg   17.90
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).   Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).   Mkg
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).   1kg   17.90
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).  Benzyl chloride, 97% [100-44-7] (α-chlorotoluene) C₄H₄CH₂CI
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).   1kg   17.90   18.20   17.90   18.20   17.90   18.20   17.90   18.20   1
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).  32,016-1    Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).    Benzyl chloride, 97% [100-44-7] (a-chlorotoluene) C₄H₄CH₂Cl
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).   1kg   17.90
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).  32,016-1  Benzyl chloride, 97% [100-44-7] (a-chlorotoluene) C₅H₅CH₂CI
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).   1kg   17.90
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).   1kg   17.90
Mutagen, possible carcinogen. Proc. Nat. Acad. Sci. U.S., 72, 979 (1975).  32,016-1  Benzyl chloride, 97% [100-44-7] (a-chlorotoluene) C₅H₅CH₂CI

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35,958-0 (8S,9R)-(·)-N-Benzylcinchonidi mp 210°(dec.) [a]8 180°(c= N-Benzylcinchoninium chioric Beil. 23,436 Fieser 12,380 1 36,618-8 ~ OBD Remainder N-benzyldihydrocinc 23,421-4 Benzyl cinnamate, 99% [103mp 37-39° bp 195-200°/5mm Index 11,1144 NMR 2(2),276C Benzyl cyanide, 99 + % [140-2 mp -24° bp 233-234° n5 1.5: Index 11,1145 NMR 2(2),395A 18,572-8 RTECS# AM1400000 Disp. A B1,940-1 Benzyl cyanide, 98% [140-29-4 32,058-7 Benzyl cyanoformate, 97% [5 $\xi$ bp 66-67°/0.6mm ng 1.5046 c Safety 2,384D Disp. A HIGHI 18,717-8 1-Benzyl-4-cyano-4-hydroxypipe (1-benzyl-4-hydroxyisonipeco FT-IR 1(2),4380 Safety 2,385A 27,798-3 N-Benzylcyclopropanecarboxar. FW 175.23 mp 142-144° Disp. B1,980-0 S-Benzyl-L-cysteine, 97% [3054 mp 214° (dec.) [a]8-10° (c = 2, 1) FT-/R 1(2),251C Disp. A 85,866-8 S-Benzyl-L-cysteine-4-nitroanilid

C-H<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH(NH<sub>4</sub>)CONHC<sub>4</sub>H<sub>4</sub>N
[a]B -68.6° (c = 0.5, dioxane) F7-IF Substrate for cystine aminopeptid: 34,525-3 S-Benzyl-L-cysteinol, 97% [8824. C,H,CH,SCH,CH(NH,)CH,OH FW [α]B -49° (c = 1.4, C,H,OH) Bell. 6( 15,544-6 Benzyl diethyl phosphite [2768-3 bp 110°/2mm ng 1.4930 d 1.4 FT-IR 1(2),552A Safety 2,385D 34,853-8
Benzyldiethyl(2,6-xylylcarbamoyli [3734-33-6] (denatonium benzo C.H.CH,N(C,H.),CH,CONHC,H.(CH, Index 11,2877 RTECS# BO665000 Benzyldimethylamine, see N,N-Di 36,822-9 Benzyl N, N-dimethyldithiocarbam FW 211.35 mp 39-41° Fp >230° Benzyldimethyldodecylamonlur. 28,088-7 C<sub>4</sub>H<sub>2</sub>CH<sub>2</sub>N[(CH<sub>2</sub>),(CH<sub>3</sub>),Br FW: Fleser 7,16 Safety 2,386A RTEC Contains - 3% water 12,719-1 N'-Benzyl-N,N-dimethylethylenedia

\* C<sub>4</sub>H<sub>2</sub>CH<sub>1</sub>NHCH<sub>2</sub>CH<sub>2</sub>N(CH<sub>2</sub>)<sub>2</sub> FW 178
Fp > 230°F(110°C) NMR 2(1),10710 IRRITANT 34,899-6 2'-Benzyl-2,2-dimethylpropionaniild (CH,),CCONHC,H,CH,C,H, FW 267. Reagent for the titration of organolith N-Benzyl-N',N'-dimethyl-N-(2-pyridy 28,738-5, Tripelennamine hydroci 86,203-7

Benzyl S-(4,6-dimethylpyrimidin-2-yl FW 274.34 mp 64-66° Disp. A NH,-protecting reagent for amino acid Chem. Eng. News, 54(22), 3 (1976). Ibid

				1019		Pina 🔳
enyi)	1g 5g		8-7	(15,25,3R,55)(+)-Pinanediol, 99% [18680-27-8] {[15-(1\alpha,2\alpha,3\alpha,5\alpha)]-2,6,6	1g - 5g	\$ 13.70 45.60
Merck	5g	4		Fp > 230°F(110°C) [α]B + 8.5° (c = 6.5, C <sub>4</sub> H <sub>4</sub> CH <sub>4</sub> ) Bell. 6(3),4145 FT-IR 1(3),250D Disp. A		
A	. "	45.2	20	Chiral reagent in the synthesis of (2S,3S)- and (2R,3S)-3-phenyl-2-butanol. <i>J. Am. Chem. Soc.</i> , <b>102</b> , 7590 (1980).		
			10-6	3.Pinanol, see isopinocampheol Pindolol, 97% [13523-86-9] {1-(1H-Indoi-4-yloxy)-3-[(1-methylethyl)amino]-2	250mg	11.80
		, A	St	propanol} FW 248.33 mp 167-171° Merck Index 11,7412 Safety 2,2840C RTECS# UB6660000 Disp. A TOXIC IRRITANT	1g	26.90
:	5g 10g	49.4	107-0	(1R)-(+)-α-Pinene, 99 + % [7785-70-8] FW 136.24 mp 62° bp 155-156°	5g 25g	17.00 55.40
		81.80	V.	Index 11,7414 Safety 2,2841A Disp. D FLAMMABLE LIQUID IRRITANT Precursor to monoisopinocampheylborane (IPCBH <sub>2</sub> ) used to prepare (+)-trans-2-		
46	20mi	93.00	350	methylcyclohexanol from 1-methylcyclohexene in 72.4% e.e. <i>J. Am. Chem. Soc.</i> , 99, 5514 (1977). Chiral intermediate. <i>Aldrichimica Acta</i> , 13(1), 13 (1980). <i>Ibid.</i> , 20(1), 24		
		. 28	8	(1987). 98 + % e.e.		*
*********	25g	20.50	688-0	(1 <i>R</i> )-(+)-α- <b>Pinene</b> , 98% [7785-70-8] [α] <sup>12</sup> + 47.1° (neat)	100mi 500ml	22.40 77.00
	100g	58.20	1111845-1	(1R)(+)-c-Pinene, tech., 85% [7785-70-8] [a]B + 43° (neat)	100ml 500ml	15.20 48.40
im	-	1	14762-4	(±)-α-Pinene, 98% [2437-95-8] FW 136.24 bp 155-156° ng 1.4650 d 0.858	5ml	11.90
	5g 25g	21.80 70.80	19.F.	Fp 90°F(32°C) Bell. 5,144 Merck Index 11,7414 NMR 2(1),52D FT-IR 1(3),75B Safety 2,2840D Disp. D FLAMMABLE LIQUID IRRITANT	250ml 1L	22.80 62.20
	. 5g	16.60	20,571-5	(1S)-(-)-\arphi-Pinene, 99% [7785-26-4] FW 136.24 bp 155-156\alpha ng 1.4650 d 0.855 Fp 90\displays(32\displays) [\alpha]g -50.7\displays(neat) Bell. 5,144 Merck Index 11,7414	5g 25g	17.40 58.50
<i>erck</i> A	100g 500g	49.75 167.50	25.10	Safety 2,2841B Disp. D FLAMMABLE LIQUID IRRITANT 98+% e.e.		. , ;
:******	5mi	10.95	₹7,39-9	(15)-(-)	25g 100g	12.70 32.60
	100mi 500mi	15.55 15 64.55	P4,570-2	(1S)-(-)-α-Pinene, 98% [7785-26-4] [α] <sup>22</sup> -42° (neat)	100ml	9.95
	,	3	11,208-9	81+% e.e. (1S)-(-)-β-Pinene, 99% [18172-67-3] FW 136.24 mp -61° bp 165-167° ng 1.4780.		29.85 11.90
			± 1.25. ★	d 0.859 Fp 91°F(32°C) [\alpha]6-21° (neat) Bell. 5,154 Merck Index 11,7415  NMR 2(1),54A FT-IR 1(3),75D Safety 2,2841C RTECS# DT5077000 Disp. D	250ml 1L	18.30 56.40
		4	00.7	FLAMMABLE LIQUID IRRITANT Chiral intermediate. Aldrichimica Acta, 13(1), 13 (1980).	. 4L	97.85
	1g 5g	17.90 65.40	21,830-8 4 502.0	α-Pinene oxide, 98% [1686-14-2] FW 152.24 bp 102-103°/50mm n̄ξ 1.4690d 0.964 Fp 150°F(65°C) [α] <sup>28</sup> -81° (neat) Bell. 5,152 · NMR 2(1),198D FT-IR 1(3),312D Safety 2,2841D RTECS# RP5600000 Disp. C	50g 250g	18.50 60.00
********	1g	27.75	21,831-6 28. *	β-Pinene oxide, 90% [6931-54-0] FW 152.24 bp 98-100°/27mm ng 1.4770 d 0.976 Fp 151°F(66°C) [α]β + 7° (neat) Beil. 17(2),44 NMR 2(1),199A FT-IR 1(1),237B Safety 2,2842A RTECS# TK4570000 Disp. C	50 <b>g</b>	26.00
3 <b>c.)</b>	1g 5g	21.60 61.00	11,010-8	c/s-Pinonic acid, 98% [473-72-3] (c/s-3-acetyl-2,2-dimethyl- cyclobutaneacetic acid) CH <sub>2</sub> COC <sub>2</sub> H <sub>4</sub> (CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub> CO <sub>2</sub> H FW 184.24 mp 104-107°	5g 25g	22.45 73.10
			ge.	Bell. 10,622 NMR 2(1),469D FT-IR 1(1),533A Safety 2,2842B Disp. C		
	1g	13.70	er	Pipecolic acid, see P4585-0, Pipecolinic acid page 1019 Pipecoline, see Methylpiperidine	٠	
145	5g	45.60	26,806-2	p-Pipecolinic acid, 99% [1723-00-8] [(R)-(+)-2-piperidinecarboxylic acid]	25mg 100mg	22.10 59.85
			P4,585-0	pL-Pipecolinic acid, 98% [4043-87-2] (2-piperidinecarboxylic acid) FW 129.16	25g	<b>41.30</b>
~~~	~	1	7	mp 282° (dec.) Beil. 22,7 Merck Index 11,7425 NMR 2(1),505C FT-IR 1(1),585B Salety 2,2843A RTECS# TK6021000, Disp. A IRRITANT	100g	112.00
Ĥ	Q Land		24,852-5	pt-Pipecolinic acid hydrochloride, 99% [5107-10-8] (2-piperidinecarboxylic acid) FW 165.62 mp 263-266° Bell. 22,7 Merck Index 11,7425	. 5g	28.95
<b>~</b>	<b>~</b>		23,775-2	FT-IR 1(1),585C Safety 2,2843B Disp. A IRRITANT HYGROSCOPIC L-Pipecolinic acid, 99% [3105-95-1] [(S)-(-)-2-piperidinecarboxylic acid]	. 100ma	23.60
				FW 129.16 mp 272° (α) <sup>23</sup> -26.4° (c = 1, H <sub>2</sub> O) Beil. 22,8 Merck Index 11,7425 NMR 2(1),505D FT-IR 1(1),585D Safety 2,2842C Disp. A IRRITANT Proline homolog. Occurs in seeds, malt, edible mushrooms, fruits, etc.		
90-0	_	·		он сн		^
			CH3 VEOI	CH3 CH2 CH3, O	(	<u>]</u>
CH2CH	ر اع دا-		<b>(</b> \(\forall \)		`	H C-OH
-			28,236	-7 28,410-6 P4,568-0 11,208-9 21,830-8 21,831-	6 <b>F</b>	P4,585-0

				1057 <sup>-</sup>	-	Pregi 🖪
	40		702-4	5β-Pregnane-3α,17α,20α-triol, 98% [1098-45-9] FW 336.52 mp 250-252°	25mg	\$ 12.40
***************************************	10g 50g	27.20	24	[a] <sup>23</sup> -3.5° (c = 0.5, C,H <sub>2</sub> OH) Bell. 6(3),6405 Safety 2,2926D Disp. A	100mg	31.90
³r FW 140.91 Merck	5g	97.60 24.9	44,768-4	Pregnencione, 98% [145-13-1] FW 316.49 mp 190-192°	5g 25g	11.20 35.60
בוב	25g	કો જે	20	FT-IR 1(2),1051D Safety 2,2940C RTECS# TU5560700 Disp. A		9.45
	4001	. 4	990-2	Pregnenoione acetate, 99% [1778-02-5] FW 358.52 mp 149-152°	25g	33.30
1.010 Fp none	500mi	14.00 🕏	Z A	FT-IR 1(2),1061C Safety 2,2940D Disp. A		
		38.80	7.7	Prehnitene, see 15,360-5, 1,2,3,4-Tetramethylbenzene page 1171		
none Disp. H	100ml	78.10		Prenyl bromide, see 24,990-4 4-Bromo-2-methyl-2-butene page 198		
oncentration	250ml	125.40		Pr(fod), see 16,135-7, Resolve-Al PrFOD® page 1088 Pr(hfc), see Tris[3-(heptafluoropropylhydroxymethylene)camphorato],		
12O FW 318.04	25a	4.0		praseodymium(iii) derivative	4	7.10
70 111 010.04	100g	18.10 51.50	8,039-3	Primaquine diphosphate, 99% [63-45-6] [8-(4-amino-1-methylbutylamino)-6 methoxyquinoline] FW 455.35 mp 205-206° (dec.) Merck index 11,7751	1g 10g	35.55
8] PrCI, 7H,O	5g	48.20	38.	NMR 2(2),741A FT-IR 1(2),864B Safety 2,2942C RTECS# VA9660000 Disp. A		
. Q IRRITANT			#.O	TOXIC	25~	12 60
PrCI,-6H,O	50g	32 00	6,686-5	Primuline [8064-60-6] (C.I. 49000, Direct Yeilow 59) FW 475.55 λmax 356nm FT-IR 1(2),1039B UV-VIs 588 RTECS# TV1050000 Disp. A	25g 100g	13.60 42.10
5000 Disp. Q	250g	32.40 122.50	68:	Useful in a simple retrograde double-labeling procedure for studying axonal		,
	٠. •			branching, in combination with 4',6-diamidino-2-phenylindole dihydrochloride (DAPI,		
77.01 B-NOV.01.0	F		72	21,708-5) and Evans Blue (20,633-4). <i>Science</i> , <b>204</b> , 873 (1979). Dye content ~75%		
77-0] Pr(NO <sub>3</sub> ) <sub>3</sub> -6H <sub>2</sub> O i400000 Disp. Q	.5g .25g	42.65 168.30	48	Pristane, see T2280-2, 2,6,10,14-Tetramethylpentadecane page 1175	5	
	·		22,296-8	Procainamide hydrochloride, 99% [614-39-1] [4-amino-N-(2	25g	8.80
		^	. 3°4 ★	dlethylaminoethyl)benzamide] H,NC,H,CONHCH,CH,N(C,H,),·HCl FW 271.79		
7] Pr(NO <sub>3</sub> ) <sub>3</sub> -6H <sub>2</sub> O	50g 250g	26.80	<b>6</b>	mp 167-169° Bell. 14(3),1077 Merck Index 11,7762 NMR 2(2),350C FT-IR 1(2),373B Safety 2,2942D RTECS# CV2295000 Disp. A IRRITANT		
'f <sub>2</sub> (C <sub>2</sub> O <sub>4</sub> ) <sub>3</sub> -xH <sub>2</sub> O	. •	97.25	94 303-5	Procaine, 99 + % [59-46-1] [2-(diethylamino)ethyl 4-aminobenzoate]	25g	11.90
-3/0304/3 ×1130	100g	17.10 48.40	1,000		100g	31.60
FW 1021.44 Disp. O.	2g	19.50	34	Index 11,7763 NMR 2(2),288B FT-IR 1(2),303C Safety 2,2943A		
•	10g	74.70	3.0	RTECS# DG2100000 Disp. A TOXIC IRRITANT  Proceins bydrochlodde 99% [51,05,9] [2,(disthylamino)ethyl	: 5g	8.10
***************************************	50g 250g	39.40	¥2,291.6	Proceine hydrochloride, 99% [51-05-8] [2-(diethylamino)ethyl4-aminobenzoate] H <sub>2</sub> NC <sub>4</sub> H <sub>4</sub> CO <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> N(C <sub>2</sub> H <sub>2</sub> ) <sub>2</sub> +CI FW 272.78 mp 155-156°	100g	14.85
FW 570.00 Merck	_	118.00 27.20	2	Bell. 14,424 NMR 2(2),349D FT-IR 1(2),303D Safety 2,2943B	_	
			<b>6</b> 8	RTECS# DG2275000 Disp. A TOXIC IRRITANT		
:00 d 0.868	782g	13.00	87,255-2	Procion Yellow H-E3G [59112-78-6] (Reactive Yellow 81) Disp. A	10g	10.00
,	6x782g 3.1kg	73.50 36.95	37,255-2	Procion reliew 11-250 [03/72-70-0] (Nodeline reliew of blop. All.	.50g	33.50
4	x3.1kg	139.00		Proflavine hemisulfate, see 19,822-6, 3,6-Diaminoacridine hemisulfate		
	15.6kg	173.20	03.	page 381		
ab	4		-10/	Proflavine hydrochloride, see 13,110-5, 3,6-Diaminoacridine hydrochloride page 381		
-chromene) F(110°C) Merck	1g 5g	29.80 106.60	85,045-4		5g	9.30
A	-g	.00.00	' -	$[\alpha]^{23}$ + 182° (c = 2, dioxane) Merck Index 11,7783 FT-IR 1(2),1052B	25g	36.20
hich induce			01.// ^ 04.//	Safety 2,2944C RTECS# TW0175000 Disp. A CANCER SUSPECT AGENT		
tial insecticides. .'s, <b>54</b> (16), 19 (1976).				MUTAGEN  L-Prolinamide, 98% [7531-52-4] FW 114.15 mp 95-97° [ $\alpha$ ] $\beta$ -100°(c = 2, C <sub>2</sub> H <sub>8</sub> OH)	250ma	11.80
-3-chromene)	250mg	21.60	20,703.3	Beil. 22(3),15 Disp. A	1g	31.80
1,7716	1ğ	53.30	85,891.9	<b>p-Proline, 99 + % [344-25-2] [(R)-(+)-proline]</b> FW 115.13 mp 223° (dec.)	100mg	6.30
hodoros parkeri.			Gra ±	[a] <sup>22</sup> + 85.0° (c = 4, H <sub>2</sub> O) Beil. 22,2 Fleser 7,307 9,393 FT-IR 1(1),583B	500mg	17.90 77.10
moderos parkom			G7182.4	Salety 2,2946D Disp. A pt-Proline, 99% [609-36-9] FW 115.13 mp 208°(dec.) Bell. 22,4	5g 1g	6.40
a-1,4-dlene-3,20	<u>1g</u>	9.10	(i,102.4 3 ★	Fleser 9,393 Merck Index 11,7790 NMR 2(1),504A FT-IR 1(1),583A	5g	19.85
ne) <i>Beil.</i> 8(4),3467 Disp. A	5g	37.40		Salety 2,2946C Disp. A		
iene-3,11,20-trione)	1g	9.85	13,154-7	L-Proline, 99 + % [147-85-3] [(S)-(·)-proline] FW 115.13 mp 228° (dec.)		5.90 10.55
8(4),3531 Merck		37.60	© × 0.	[a]B-84° (c = 4, H,O) Bell. 22,2 Fieser 6,492 8,421 9,393 10,331 12,414 Merck Index 11,7790 FT-IR 1(1),583C Safety 2,2946A RTECS# TW3584000 Disp. A	100g	33.50
Α		2	O:	Optically active intermediate for organic synthesis. Aldrichimica Acta, 13(1), 13	•	
esterone page 375		45.00		(1980).		
α,20α-dlol)	1mg 5mg	15.80 50.45	Ì		•	
√-5β-pregnane)	100ma	17.80	сн₃о√	^ ^		
ck Index 11,7732	500mg	61.55	01.30 X	· 2H <sub>3</sub> PO <sub>4</sub>		٦.
TU4157113 Disp. A				CH S S S S S S S S S S S S S S S S S S S	\ <sub>N</sub> .	У∭-он
CH30			1	NHCH(CH <sub>2</sub> ) <sub>3</sub> NH <sub>2</sub> so <sub>3</sub> Na so <sub>1</sub> NH	Ĥ	
CH3O	3			CH <sub>3</sub> NH <sub>2</sub>	40	154-7
19,491-3				UEST AVAILABLE COPY 28,705-9	13,	154-7
				CT AVAILABLE OU .		
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# Fundamentals of Organic Chemistry

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Geo:

Acid halides are distinguished from the corresponding acids by a carbonyl stretching band at a significantly higher frequency in the infrared. Whereas type AC carboxylic acids absorb at 1725–1700 cm<sup>-1</sup>, the corresponding acid chlorides absorb at 1815–1770 cm<sup>-1</sup> (see Table 15.2 on p. 394).

15.1c. Synthesis of acid halides. Acid halides are almost invariably synthesized from the corresponding carboxylic acids by the action of any one of several inorganic reagents, including phosphorus trichloride (PCl<sub>3</sub>), phosphorus pentachloride (PCl<sub>5</sub>), and thionyl chloride (SOCl<sub>2</sub>) (Fig. 15.1).

$$CH_3C \stackrel{O}{\longrightarrow} + PCI_3 \longrightarrow CH_3C \stackrel{O}{\longrightarrow} + H_3PO_3$$

$$CH_3CH_2CH_2C \stackrel{O}{\longrightarrow} + SOCI_2 \longrightarrow CH_3CH_2CH_2C \stackrel{O}{\longrightarrow} + SO_2 + HCI$$

Fig. 15.1. Synthesis of acid chlorides from carboxylic acids.

15.1d. REACTIONS OF ACID HALIDES. We might predict that the electrophilic character of the carbonyl carbon in acid halides should be greater than in aldehydes and ketones because of the electron-withdrawing effect (i.e., inductive effect) of the halogen atom. This does, in fact, prove to be the case, although to some extent this effect is counterbalanced by the electron-releasing effect of the nonbonded electrons of the halogen [e.g., resonance structure Fig. 15.2(c)];

Fig. 15.2. Resonance structures of acid chlorides.

acid halides are exceedingly reactive toward nucleophilic reagents. The products of reaction, however, are different from those from aldehydes and ketones (see Section 13.5), for the initial step, producing an addition product, is succeeded by a second step, in which the halogen is eliminated and the carbonyl group is regenerated. The overall reaction is a *substitution* process, which proceeds via nucleophilic addition followed by elimination. The hydrolysis of acetyl chloride, for instance, can be depicted in this fashion (Fig. 15.3).



First, amino acids are bifunctional molecules capable of undergoing facile reaction at the carboxyl function as well as the amino function. Thus, if we wish to form a peptide link between two different amino acids, we immediately face the problem associated with any mixed condensation—viz., the formation of mixtures of products:

Chapter 39
Organic Synthesis

To circumvent this problem, we must "protect" or "block" the amino group of one of the participants and the carboxyl group of the other. Amino groups are frequently protected by means of the t-butoxycarbonyl (t-BOC) function, prepared by the action of t-butyl azidoformate  $[(CH_3)_3COCON_3]$  or t-butyl chloroformate  $[(CH_3)_3COCOCl]$  on the amino acid. Carboxyl groups can be protected by conversion to the benzyl or t-butyl ester. All of these are good protecting groups because they are easily removed under mild acid-catalyzed hydrolysis (Fig. 39.9).

$$H_2NCHCO_2H + OCH_2OCCO_NHCHCO_2H$$
 $H_2NCHCO_2H + HOCH_2 OCCO_NHCHCO_2CH_2 OCCO_NHCHCO_2CH_2 OCCO_NHCHCO_2CH_2 OCCO_NHCHCNHCHCO_2CH_2 OCCO_NHCHCNHCHCO_2CH_2 OCCO_NHCHCNHCHCO_2CH_2 OCCO_NHCHCNHCHCO_2CH_2 OCCO_NHCHCNHCHCO_2CH_2 OCCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO_NHCHCO$ 

Fig. 39.9. Use of "protecting" groups in the synthesis of a dipeptide.